



Lithiation of 1*H*-Pyrazolo[3,4-*d*]pyrimidine Derivative Using Lithium Alkanetellurolate

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Abs tract: 4-Chloro-1-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidine was converted into alkyl 1-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidin-4-yl telluride, which was lithiated using alkyllithium followed by the reaction with electrophiles. ⊚ 1999 Elsevier Science Ltd. All rights reserved.

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The halogen-metal exchange reaction is widely used for the introduction of electrophiles in aromatic compounds. In some π -deficient heterocycles, treatment with an organolithium reagent gives multiproducts derived from some side reactions. Very low temperature is required for the metallation of π -deficient heterocycles. For example, it has been reported that 7-iodo-3-phenyl-3*H*-1,2,3-triazolo[4,5-*d*]pyrimidine is lithiated at the 7-position [1], but this reaction requires a temperature of -105 °C, and produces some byproducts. Similarly, the reaction of 4-iodo-1-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidine with *n*-butyllithium in THF at -78 °C gave multispots in Thin Layer Chromatography. We examined the lithiation of π -deficient heterocycles under mild conditions.

Recently, lithiation via a tellurium-lithium exchange reaction has been reported [2,3,4]. As regards π -deficient heterocycles, the lithiation of 2-bromopyridine using n-butyllithium via n-butyl 2-pyridyl telluride was reported by Kondo [5]. This method can avoid some side reactions which occur due to the superiority of the halogeno group as a leaving group. Thus, we applied the method to the lithiation of 4-chloro-1-phenyl-1H-pyrazolo [3,4-d] pyrimidine.

Reaction of 4-chloro-1-phenyl-1H-pyrazolo[3,4-d]pyrimidine (1) with lithium n-butanetellurolate, which was obtained from the reaction of tellurium and n-butyllithium, proceeded smoothly to give n-butyl 1-phenyl-1H-pyrazolo[3,4-d]pyrimidin-4-yl telluride (2) in 90% yield. Then 2 was converted into the product 4 using n-butyllithium and pivalaldehyde

in 52 % yield. Because the telluride 2 is slightly unstable, conversion of 2 into 4 does not proceed in good yield.

Next, the results of the one-pot lithiation without isolating the telluride 2 are shown in Table 1. When the lithiating time was extended to 90 min, the yield of the product 4 showed a slight decline. Even at -78 °C, hence, 4-lithio derivative 3 seems to be unstable. Upon using methyllithium or phenyllithium instead of n-butyllithium, we found that the kind of alkyllithium affects the yield of the product 4. It should be noted that treatment of 2, generated in situ from 1, with excess molar of n-butyllithium gives the product 5, which is derived from the nucleophilic attack of n-butyllithium at the 6-position of the 4-lithio derivative 3 (entry 3 in Table 1). The introduction of some electrophiles at the 4-position in 1-phenyl-1H-pyrazolo[3,4-d]pyrimidine was accomplished in good to fair yields under the best conditions we examined.

Table 1

reagents and conditions: i) RTeLi (1.1 eq) / THF / rt; ii) RLi / -78 °C; iii) electrophile (5.0 eq) / -78 °C to rt; iv) H₃O⁺ / rt

	Alkyllithium (RLi)					Yield	
Entry	R	Amount	Time	Electrophile	-E	4	5
1	ⁿ Bu	1.1 eq	10 min	^t BuCH=O	-CH(OH) ^t Bu	74 %	
2	ⁿ Bu	1.1 eq	90 min	^t BuCH=O	-CH(OH) ^t Bu	50 %	
3	ⁿ Bu	3.0 eq	10 min	^t BuCH≕O	-CH(OH) ^t Bu	43 %	14 %
4	Me	1.1 eq	10 min	^t BuCH≃O	-CH(OH) ^t Bu	48 %	
5	Ph	1.1 eq	10 min	^t BuCH=O	-CH(OH) ^t Bu	16 %	
6	ⁿ Bu	1.1 eq	10 min	PhCH=O	-CH(OH)Ph	61 %	
7	ⁿ Bu	1.1 eq	10 min	Me ₂ NCH=O	-CH=O	60 %	

In conclusion, we have accomplished the lithiation of pyrazolo[3,4-d]pyrimidine derivative using lithium alkanetellurolate. We hope that this method can be applied to other π -deficient heterocycles.

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